

DEPT. DISPATCHED TO ABOVE ADDRESS ON AN AMONIA LEAK IN REFRIDGERATION SYSTEM, APON ARRIVAL FOUND REFRIDGERATION PUMP ROOM FULL OF ANHYDRUS AMONIA VAPERS, POSITIVE PRESSUR VENTILATION WAS STARTED AND VALVE SHUT DOWN PROCEDURES STARTED, IT WAS LEARNED FROM MAINT. MAN THAT SYSTEM HAS A 1500 LB. CAPACITY ALL VALVES WERE CLOSED EXCEPT MAIN FEED VALVE WICH WAS ICED OVER IT WAS NECESSARY TO CHIP ICE FROM PIPE RUN IN ORDER TO CLOSE VALVE, THIS PROCEDURE WAS HAMPERED DUE TO AMONIA VAPERS PENETRATING FIREFIGHTERS BUNKER GEAR CAUSING SEVERE BURNING SENSATION. WORK WAS DONE IN TWO TO THREE MIN. SHIFTS BY SIX FIREFIGHTERS WHO LATER WERE TRANSPORTED TO E.R. REFRIDGERATION TECK. RON EXLINE OF REFRIDGERATION SYSTEMS INC. COLS. OHIO ARRIVED AND CHECKED UNIT AFTER LEAK WAS STOPED AND STATED HE WOULD NOT KNOW THE AMOUNT OF PRODUCT THAT WAS LOST UNTIL HE GETS SYSTON BACK ON LINE BUT ESTAMATED 200 TO 300 LBS.HE SEAD HE WOULD FORWARD ACCURATE ASSESMENT TO F.P. OFFICE LATER FIREFIGHTERS TRANSPORTED TO E.R. --- LT. A. BEVINGER TREATED FOR RESPIRATORY AND EXPOSURE TO VAPERS, LT. H. SMITH, LT. M. LEHMAN, FFP. R. HOFFMAN F.F. H. BARRERA , M. DAWSON TREATED FOR EXPOSURE TO VAPERS ALL LATER RELEASED.

NOTE: JUDY FARMER CALLED LESS THAN 100 LBS. RELEASED.



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OCTOBER 19, 1993

MR. DAN BOLGER
FAIRFIELD COUNTY LEPC
BOX 626
LANCASTER, OHIO 43130

DEAR DAN,

ENCLOSED IS A COPY OF THE SERVICE REPORT FROM REFRIGERATION SYSTEMS FOR THE AMMONIA LEAK ON THURSDAY, OCTOBER 14, 1993. THE ESTIMATED QUANTITY OF LOSS WAS LESS THAN 100 LB.

SINCERELY YOURS,

A handwritten signature in blue ink that reads "Julia Farmer".

JULIA FARMER
PRESIDENT

ENCLOSURE

Refrigeration Systems Company
 770 Genessee Avenue - Columbus, Ohio 43211
 Ph. (614) 263-0913 Fax (614) 263-6660

- CINCINNATI OFFICE (513) 793-4463
- CLEVELAND OFFICE (216) 562-8074
- EASTON, PA OFFICE (215) 252-6990

SERVICE OR DELIVERY LOCATION

NAME
 STREET
 CITY
 STATE
 ZIP
 CONTACT NAME
 CONTACT PHONE NO.

BILLING ADDRESS - IF DIFFERENT

NAME
 STREET
 CITY
 STATE
 ZIP
 CONTACT PHONE NO.

SERVICE REPORT NO

52994

DATE RECEIVED
 10-15-93

TECHNICIAN
 Ken Exlar

EMERGENCY
 YES
 NO

PARTS ONLY
 SERVICE CONTRACT
 TIME AND MATERIAL A JOB

UNIT PRICE
 AMOUNT

WORK TO BE DONE:

NI3 Leak in comp

MODEL:

SERIAL NUMBER:

DESCRIPTION OF WORK PERFORMED

The 3/8 line on large NI3 pump was leaking by ammonia. The gas Dept. had pump replaced by the time I got here. I should check valve, put charge hose on pump & purge into water bucket. Restarted system & checked for proper operation.
 Total loss of NI3 is less than 100 lb.

REGULAR HOURS JOURNEYMEN	REGULAR HOURS APPRENTICE	OVERTIME HOURS JOURNEYMEN	OVERTIME HOURS APPRENTICE	HOURS	RATE	TOTAL
			4			
TOTAL MATERIAL						
TOTAL LABOR						
OTHER						
SALES TAX						

EXPENSES:

TRAVEL:

SHIPPING CHARGE:

I HEREBY ACKNOWLEDGE THE SATISFACTORY COMPLETION OF THE ABOVE DESCRIBED WORK.

X *Paul A. Witt* CUSTOMER SIGNATURE

DATE

SERVICE REPORT

TOTAL CHARGE

OR FIRE INCIDENT REPORTING SYSTEM

INCIDENT REPORT

Lancaster Fire Department

A	FDID	INCIDENT NO.	EXP	DATE	DAY OF WEEK	ALARM	ARRIVAL	IN SERVICE
	23011	000752	00	10/14/93	Thursday	5	21:16	21:18 24:00
B	SITUATION FOUND							
	Spill, Leak w/ No Ign 41							
C	ACTION TAKEN				MUTUAL AID			
	Remove Hazard 4				<input type="checkbox"/>			
D	FIXED PROPERTY USE				IGNITION FACTOR			
	Food, Sales Not Class 519				NOT A FIRE 00			
E	CORRECT ADDRESS						ZIP CODE	CENSUS TRACT
	505 W. FAIR AVE						43130	
F	OCCUPANT NAME				TELEPHONE		ROOM OR APT.	
	BETRY LANCASTER				(614) 653-1052			
G	OWNER NAME			ADDRESS			TELEPHONE	
	BROWN HELEN							
H	METHOD OF ALARM		CO. INSPECTION DISTRICT		SHIFT		NO. ALARMS	
	Telephone (Direct) 1		2		1		1	
I	NO. FIRE PERSONNEL RESPONDED		NO. ENGINES RESPONDED		NO. AERIAL APPARATUS RESPONDED		NO. OTHER VEHICLES RESPONDED	
	15		2		1		2	
J	NUMBER OF INJURIES		OTHER		NO. OF FATALITIES		OTHERS	
	FIRE SERVICE 6		0		FIRE SERVICE 0		0	
K	COMPLEX				MOBILE PROPERTY TYPE			
L	AREA OF FIRE ORIGIN				EQUIPMENT INVOLVED IN IGNITION			
M	FORM OF HEAT IGNITION		TYPE OF MATERIAL IGHITED		FORM OF MATERIAL IGHITED			
N	METHOD OF EXTINGUISHMENT			LEVEL OF FIRE ORIGIN			ESTIMATED TOTAL DOLLAR LOSS	
	<input type="checkbox"/>			<input type="checkbox"/>			1,500.00 0	
O	NUMBER OF STORIES				CONSTRUCTION TYPE			
	<input type="checkbox"/>				<input type="checkbox"/>			
P	EXTENT OF DAMAGE		DETECTOR PERFORMANCE		SPRINKLER PERFORMANCE			
	Flame <input type="checkbox"/> Smoke <input type="checkbox"/>		P <input type="checkbox"/>		<input type="checkbox"/>			
Q	TYPE OF MATERIAL GENERATING MOST SMOKE				AVENUE OF SMOKE TRAVEL			
					<input type="checkbox"/>			
R	FORM OF MATERIAL GENERATING MOST SMOKE							
S	MOBILE PROPERTY		YEAR	MAKE	MODEL	SERIAL NO.	LICENSE NO.	
T	EQUIP INVOLVED							
U	MEMBER MAKING REPORT				OFFICER IN CHARGE (if different)			
	DATE 00/00/00				LT. A. BEYINGER Lt. upressio 10/15/93			

REMARKS

Comments for this incident have been printed on an additional comments page.

Remarks:

NOTE INJURED FIRE FIGHTERS WERE TREATED AND RELEASED FROM E.R. FOR RESPIRATORY TRUBLE AND/OR SKIN EXPOSURE TO ANHYDROUS AMONIA VAPORS. DUE TO VAPORS PERMINATING BUNKER GEAR. DUE TO HEAVY CONCENTRATION

Trucks:	Drivers:	#Men:	Assigned Personnel	On Duty	At Scene	Masks Worn	Injuries
AP-1	UHL	3	Capt. D. Crozier				
P-6	BRIDGES	3	Lt. A. Bevinger	✓	✓	✓	✓
P-7	MOYER	3	Lt. H. Smith	✓	✓	✓	✓
M-2	BUSHNES	3	Lt. J. Kraner				
M-3	G. MILLER	3	Lt. M. Lehman	✓	✓	✓	✓
			FFE. S. Uhl	✓	✓		
			FFP. G. Miller	✓	✓		
			FFP. N. Northrup	✓	✓		
			FFE. J. Beck				
			FF. M. Harris				
			FFE. P. Miller				
# Ft. Hose:	# Ft. Ladders:	Extinguishers:	FFP. R. Hoffman	✓	✓	✓	✓
			FFE. D. Sells	✓	✓		
			FFP. B. Archer	✓	✓		
			FFP. M. Touvell				
			FF. H. Barrera	✓	✓	✓	✓
			FFP. G. Bushee	✓	✓		
			FFE. C. Moyer	✓	✓		
Approx. Gallons of Water Used:			FF. B. Knisley	✓	✓		
Other Equipment Used:			FFP. B. Squires				
			GASOLINE EXHAUST (AN)	✓	✓		
			FF. D. Comer	✓	✓		
			FF. M. Dawson	✓	✓	✓	✓
			FF. R. Howdysshell				
Damage to Apparatus or Equipment:							

=====

CHEMICAL: AMMONIA, [ANHYDROUS]

CAS #: 7664-41-7

NOAA #: 4860

UN #: 1005 STCC: 4904210

RTECS: BO0875000

FORMULA: H3N

LABEL: POISON GAS

NFPA CODES: H3 F1 R0 S

CERCLA (Y/N): Y

EHS (Y/N): Y

313 (Y/N): Y

RCRA:

RQ: 100

TPQ: 500

LAST UPDATE:10/20/92

STATE at ambient temperature: [Gas, Liquid, Solid] (G/L/S): G

LEVEL OF CONCERN: 0.03500000 gm/m3

LIQUID AMBIENT FACTOR:

LIQUID BOILING FACTOR:

LIQUID MOLTEN FACTOR:

CAMEO Response Information

[NOAA, 7600 Sand Point Way NE, Seattle, WA 98115 (206) 526-6317]

GENERAL DESCRIPTION:

Anhydrous ammonia is a clear colorless gas with a characteristic odor. It is used as a fertilizer, as a refrigerant, and in the manufacture of other chemicals. Although it is classed as a nonflammable gas, it will burn within certain vapor concentration limits, and the fire hazard will increase in the presence of oil or other combustible materials. Its "combustibility" is definitely not a common problem in the event of leakage. It is shipped as a liquid under pressure. Contact with the liquid can cause frostbite. It is soluble in water forming a corrosive liquid. Although ammonia is lighter than air, the vapors from a leak initially hug the ground. It weighs 6 lbs/gallon. Long term exposure to low concentrations or short term exposure to high concentrations can result in adverse health effects from inhalation. ((c) AAR, 1991).¹ If not breathing, provide artificial respiration. If breathing is labored, administer oxygen or other respiratory support.² DO NOT induce vomiting or attempt to

neutralize! \3. Obtain authorization and/or further instructions from the local hospital for administration of an antidote or performance of other invasive procedures.\4. Activated charcoal does not strongly bind ammonia, and therefore is of little or no value.\5. Give the victims water or milk: children up to 1 year old, 125 mL (4 oz or 1/2 cup); children 1 to 12 years old, 200 mL (6 oz or 3/4 cup); adults, 250 mL (8 oz or 1 cup). Water or milk should be given only if victims are conscious and alert.\6. Transport to a health care facility. (EPA, 1990)

FIRE & EXPLOSIVE HAZARD:

Mixing of ammonia with several chemicals can cause severe fire hazards and/or explosions. Ammonia in container may explode in heat of fire. Incompatible with many materials including silver and gold salts, halogens, alkali metals, nitrogen trichloride, potassium chlorate, chromyl chloride, oxygen halides, acid vapors, azides, ethylene oxide, picric acid and many other chemicals. Mixing with other chemicals and water. Hazardous polymerization may not occur.

(EPA, 1990)N||●

FIRE FIGHTING:

Wear positive pressure breathing apparatus and full protective clothing.

Small fires: dry chemical or carbon dioxide. Large fires: water spray, fog or foam. Apply water gently to the surface. Do not get water inside container. Move container from fire area if you can do it without risk. Stay away from ends of tanks. Cool containers that are exposed to flames with water from the side until well after fire is out. Isolate area until gas has dispersed. (EPA, 1990)R≤

PROTECTIVE CLOTHING AND SUIT MATERIAL COMPATIBILITY (ACGIN 1985:)

For emergency situations, wear a positive pressure, pressure-demand, full facepiece self-contained breathing apparatus (SCBA) or pressure-demand supplied air respirator with escape SCBA and a fully-encapsulating, chemical resistant suit. (EPA, 1990)

MATERIAL RATINGS

BARRICADE

FABRIC 1-3 hours

BUTYL

FABRIC GT 3 hours

GLOVES GT 3 hours

BUTYL/NEOP

FABRIC GT 3 hours

CHECKMATE

FABRIC GT 3 hours

CHEMREL

FABRIC LT 1 hour

CHEMREL MAX		
FABRIC	GT 3 hours	
CPE		
FABRIC	1-3 hours	
CPF III		
FABRIC	LT 1 hour	
FEP TEFLON		
FACESHIELD	GT 3 hours	
INTERCEPTOR		
FABRIC	GT 3 hours	
NAT RUB		
FABRIC	LT 1 hour	
NEOP		
FABRIC	1-3 hours	
GLOVES	GT 3 hours	
BOOTS	GT 3 hours	
NEOP/BUTYL		
GLOVES	GT 3 hours	
NIT+POLYURETHANE+PVC		
BOOTS	GT 3 hours	
NITRILE		
GLOVES	GT 3 hours	
PE		
FABRIC	LT 1 hour	
POLYURETHANE		
FABRIC	LT 1 hour	
PTFE TEFLON		
FABRIC	LT 1 hour	
PVC		
FABRIC	LT 1 hour	
RESPONDER		
FABRIC	GT 3 hours	
SARANEX23P		
FABRIC	LT 1 hour	
VITON		
FABRIC	LT 1 hour	
VITON/NEOP		
FABRIC	GT 3 hours	
♦PRINTER		

NONFIRE RESPONSE:

Keep material out of water sources and sewers. Attempt to stop leak if without undue personnel hazard. Use water spray to knock-down vapors. Vapor knockdown water is corrosive or toxic and should be diked for containment. Land spill: Dig a pit, pond, lagoon, holding area to contain liquid or solid material. Dike surface flow using soil, sand bags, foamed polyurethane, or foamed concrete. Absorb bulk liquid with fly ash or cement powder. Neutralize with vinegar or other dilute acid. Water spill: Neutralize with dilute acid. Use mechanical dredges or lifts to remove immobilized masses of pollutants and precipitates. ((c) AAR, 1991)rC Keep material out of water sources and sewers. Attempt to stop leak

HEALTH HAZARDS:

Vapors cause irritation of eyes and respiratory tract. Liquid will burn skin and eyes. Poisonous; may be fatal if inhaled. Contact may cause burns to skin and eyes. Contact with liquid may cause frostbite. (EPA, 1990)oI

FIRST AID:

Warning: Ammonia is extremely corrosive to the skin, eyes, and mucous membranes. Contact with the liquified gas may cause frostbite. Caution is advised.

Signs and Symptoms of Acute Ammonia Exposure: Inhalation of ammonia may cause irritation and burns of the respiratory tract, laryngitis, dyspnea (shortness of breath), stridor (high-pitched respirations), and chest pain. Pulmonary edema and pneumonia may also result from inhalation. A pink frothy sputum, convulsions, and coma are often seen following exposure to high concentrations. When ammonia is ingested, nausea and vomiting may result; oral, esophageal, and stomach burns are common. If ammonia has contacted the eyes, irritation, pain, conjunctivitis (red, inflamed eyes), lacrimation (tearing), and corneal erosion may occur. Loss of vision is possible.

Dermal exposure may result in severe burns and pain.

Emergency Life-Support Procedures: Acute exposure to ammonia may require decontamination and life support for the victims. Emergency personnel should wear protective clothing appropriate to the type and degree of contamination. Air-purifying or supplied-air respiratory equipment should also be worn, as necessary.

Inhalation Exposure:

1. Move victims to fresh air. Emergency personnel should avoid self-exposure to ammonia.
2. Evaluate vital signs including pulse and respiratory rate, and note any trauma. If no pulse is detected, provide CPR. If not breathing, provide artificial respiration. If breathing is labored, administer oxygen or other respiratory support.
3. Obtain authorization and/or further instructions from the local

hospital for administration of an antidote or performance of other invasive procedures.

4. Transport to a health care facility.

Dermal/Eye Exposure:

1. Remove victims from exposure. Emergency personnel should avoid self-exposure to ammonia.

2. Evaluate vital signs including pulse and respiratory rate, and note any trauma. If no pulse is detected, provide CPR. If not breathing, provide artificial respiration. If breathing is labored, administer oxygen or other respiratory support.

Warning: Do not attempt to neutralize with an acid wash; excessive liberation of heat may result.

3. If eye exposure has occurred, eyes must IMMEDIATELY be flushed with lukewarm water for at least 15 minutes.

4. Remove contaminated clothing as soon as possible.

5. Wash exposed skin areas THOROUGHLY with soap and water.

6. Obtain authorization and/or further instructions from the local hospital for administration of an antidote or performance of other invasive procedures.

7. Transport to a health care facility.

Ingestion Exposure:

1. Evaluate vital signs including pulse and respiratory rate, and note any trauma. If no pulse is detected, provide CPR. If not breathing, provide artificial respiration. If breathing is labored, administer oxygen or other respiratory support.

2. DO NOT induce vomiting or attempt to neutralize!

3. Obtain authorization and/or further instructions from the local hospital for administration of an antidote or performance of other invasive procedures.

4. Activated charcoal does not strongly bind ammonia, and therefore is of little or no value.

5. Give the victims water or milk: children up to 1 year old, 125 mL (4 oz or 1/2 cup); children 1 to 12 years old, 200 mL (6 oz or 3/4 cup); adults, 250 mL (8 oz or 1 cup). Water or milk should be given only if victims are conscious and alert.

6. Transport to a health care facility. (EPA, 1990) ¶I#♦

CHEMICAL PROPERTIES:

Flash Point: Not Applicable. Not flammable under conditions likely to be encountered. (USCG, 1991)

Lower Exp Limit: 16 % (EPA, 1990)

Upper Exp Limit: 25 % (EPA, 1990)

Auto IgnTemp: 1204 F (USCG, 1991)

Melting Point: -107.9 F (EPA, 1990)

Vapor Pressure: 400 mm at -49.72 F (EPA, 1990)

Vapor Density (air = 1): 0.6 (EPA, 1990)

Specific Gravity, Liquid: 0.6818 at -28.03 F (EPA, 1990)

Boiling Point: -28.03 F at 760 mm (EPA, 1990)

Molecular Weight: 17.03 (EPA, 1990)

IDLH: 500 ppm (NIOSH, 1990)

TLV TWA: 25 ppm ((c)ACGIH, 1991)

TLV STEL: 35 ppm ((c)ACGIH, 1991)moà